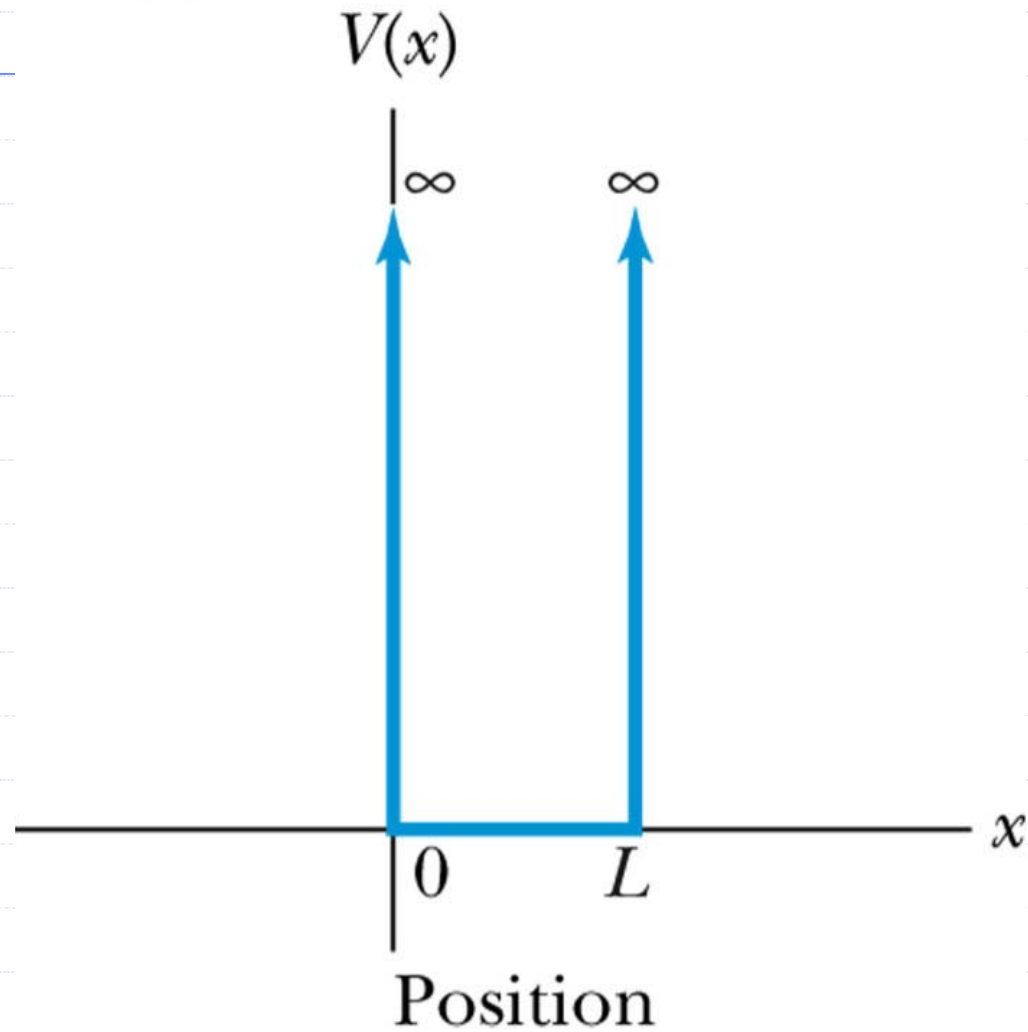


# Infinite Square Well



# Infinite Square Well

- We can use the time independent Schrodinger equation
- Outside the well

$$\psi(x) = 0$$

- Inside the well

$$-\frac{\hbar^2}{2m} \frac{d^2\psi}{dx^2} = E\psi$$

$$\frac{d^2\psi}{dx^2} = -k^2\psi \text{ where } k = \frac{\sqrt{2mE}}{\hbar}$$

Try solutions  $\psi(x) = A \sin kx + B \cos kx$

# Infinite Square Well

- Typically the constants are determined by the boundary conditions (and normalization)
- In this case the boundary conditions needed to ensure continuity of  $\psi$  are

$$\psi(0) = \psi(L) = 0$$

- Thus

$$\text{Since } \psi(0) = A \sin 0 + B \cos 0 = 0, B = 0$$

$$\text{Since } \psi(L) = A \sin kL = 0, \text{ then } kL = 0, \pm\pi, \pm2\pi, \dots$$

$$k \text{ cannot be 0, so solutions have } k_n = \frac{n\pi}{L}, n = 1, 2, 3, \dots$$

$$E_n = \frac{\hbar^2 k_n^2}{2m} = n^2 \frac{\pi^2 \hbar^2}{2mL^2}; \text{ energy is quantized!!}$$

# Infinite Square Well

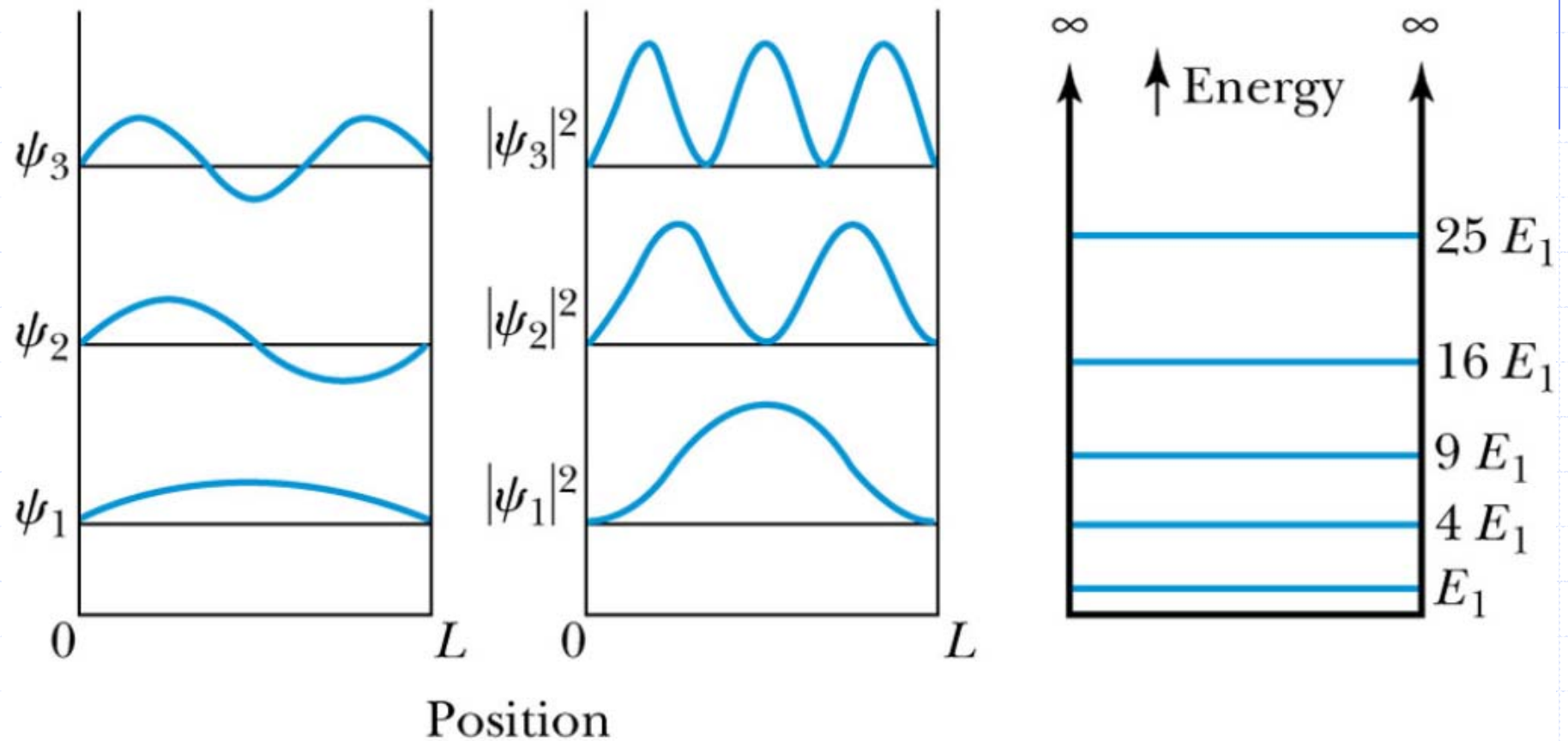
➤ What about the normalization of  $\psi_n$ ?

$$\begin{aligned}\int_0^L dx |\psi_n|^2 &= \int_0^L dx |A|^2 \sin^2 kx \\ &= |A|^2 \left( \frac{x}{2} - \frac{\sin 2kx}{4k} \right) \Bigg|_{x=0}^{x=L} \\ &= |A|^2 \frac{L}{2} = 1\end{aligned}$$

$$\text{thus } A = \sqrt{\frac{2}{L}}$$

$$\text{so the solutions are } \psi_n = \sqrt{\frac{2}{L}} \sin\left(\frac{n\pi x}{L}\right)$$

# Infinite Square Well



# Infinite Square Well

## ➤ Comments

- $\Psi$  and  $d\Psi/dx$  must be continuous at all boundaries except where the potential is infinite
- As  $n$  increases,  $\lambda$  decreases,  $E$  increases

$$k_n = \frac{n\pi}{L} = \frac{2\pi}{\lambda}$$

$$\lambda = \frac{2L}{n}$$

- Number of nodes in the  $n$ 'th eigenfunction =  $n-1$
- The eigenfunctions are alternating even and odd functions about the symmetry axis

# Infinite Square Well

## ➤ Comments

- Energy states are

$$E_n = n^2 \frac{\pi^2 \hbar^2}{2mL^2} = n^2 E_1$$

- The lowest energy bound state has non-zero energy (zero point energy)
- The excitation/de-excitation energy is  $E_{\text{final}} - E_{\text{initial}}$

# Infinite Square Well

## ➤ Comments

- The eigenfunctions  $\psi$  are mutually orthogonal (recall postulate #4)

$$\int_0^L dx \psi_i^* \psi_j = \frac{2}{L} \int_0^L dx \sin \frac{i\pi x}{L} \sin \frac{j\pi x}{L} = \delta_{ij}$$

- Please look at the calculation of expectation values in Thornton-Rex (Example 6.8)
  - ◆ What would you expect for  $\langle x \rangle?$ ,  $\langle p \rangle?$ ,  $\langle p^2 \rangle?$



# Infinite Square Well

## ➤ Comments

- What are the possible momentum states?

Look at the eigenvalue equation

$$-i\hbar \frac{\partial f_n(x)}{\partial x} = p_n f_n(x) = \pm \hbar k_n f_n(x)$$

Solutions are  $f_n(x) = Ae^{\pm ik_n x}$

Makes sense since our energy eigenfunctions can be written as

$$\sin k_n x = \frac{1}{2i} (e^{ik_n x} - e^{-ik_n x})$$

# Infinite Square Well

## ➤ Comments

- The stationary states of the infinite well are

$$\Psi_n(x, t) = \sqrt{\frac{2}{L}} \sin\left(\frac{n\pi x}{L}\right) e^{-iE_n t/\hbar} \quad \text{where } E_n = \frac{n^2 \pi^2 \hbar^2}{2mL^2}$$

- A general solution to the time dependent Schrodinger equation (see postulate #4) is

$$\Psi(x, t) = \sum_{n=1}^{\infty} c_n \sqrt{\frac{2}{L}} \sin\left(\frac{n\pi x}{L}\right) e^{-iE_n t/\hbar}$$

- Given an initial condition the coefficients  $c_n$  can be determined

$$c_n = \sqrt{\frac{2}{L}} \int_0^L dx \sin\left(\frac{n\pi x}{L}\right) \Psi(x, 0)$$

# Infinite Square Well

## ➤ Problem

■ Let

$$\psi(x) = \frac{1}{\sqrt{5L}} \sin \frac{\pi x}{L} + \frac{3}{\sqrt{5L}} \sin \frac{3\pi x}{L}$$

- What are the possible results of an energy measurement?
- What are the probabilities?
- What is the form of the wave function immediately after an energy measurement?

# Infinite Square Well

## ➤ Problem

- What is the ratio of probabilities for the particle to be at  $x=L/3$  to  $x=L/4$ ?

# Infinite Square Well

## ➤ Problem

- Let

$$\Psi(x,0) = A\{\psi_1(x) + \psi_2(x)\}$$

- Normalize  $\psi$
- Calculate  $|\psi(x,t)|^2$
- Calculate  $\langle x \rangle$
- Calculate  $\langle H \rangle$

# Infinite Square Well

## ➤ Note to myself

- [www.phys.uri.edu/~yoon/deepwellmain.html](http://www.phys.uri.edu/~yoon/deepwellmain.html)

# Infinite Square Well

➤ In three dimensions, Schrodinger's equation becomes

$$\hat{p}_x = -i\hbar \frac{\partial}{\partial x}, \hat{p}_y = -i\hbar \frac{\partial}{\partial y}, \hat{p}_z = -i\hbar \frac{\partial}{\partial z}$$

$$\vec{\hat{p}} = -i\hbar \vec{\nabla} \text{ and } \hat{p}^2 = -\hbar^2 \nabla^2$$

$$\text{where } \nabla^2 = \frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2}$$

$$i\hbar \frac{\partial \Psi}{\partial t} = -\frac{\hbar^2}{2m} \nabla^2 \Psi + V\Psi$$

$$\text{and } \Psi(x, t) \rightarrow \Psi(\vec{r}, t)$$

# Infinite Square Well

## ➤ Free particle in a box

- An infinite well in 3 dimensions

$$-\frac{\hbar^2}{2m} \nabla^2 \psi = E \psi \quad (1)$$

Try  $\psi = A \sin k_1 x \sin k_2 y \sin k_3 z$

Boundary conditions give

$$k_1 = \frac{n_1 \pi}{L_1} \quad k_2 = \frac{n_2 \pi}{L_2} \quad k_3 = \frac{n_3 \pi}{L_3}$$

Since  $\frac{\partial^2 \psi}{\partial x^2} = -k_1^2 \psi$  (1) becomes

$$\frac{\hbar^2}{2m} (k_1^2 + k_2^2 + k_3^2) \psi = E \psi$$



# Infinite Square Well

## ➤ Free particle in box

### ■ Continuing

$$E = \frac{\hbar^2}{2m} (k_1^2 + k_2^2 + k_3^2)$$

$$E = \frac{\pi^2 \hbar^2}{2m} \left( \frac{n_1^2}{L_1^2} + \frac{n_2^2}{L_2^2} + \frac{n_3^2}{L_3^2} \right)$$

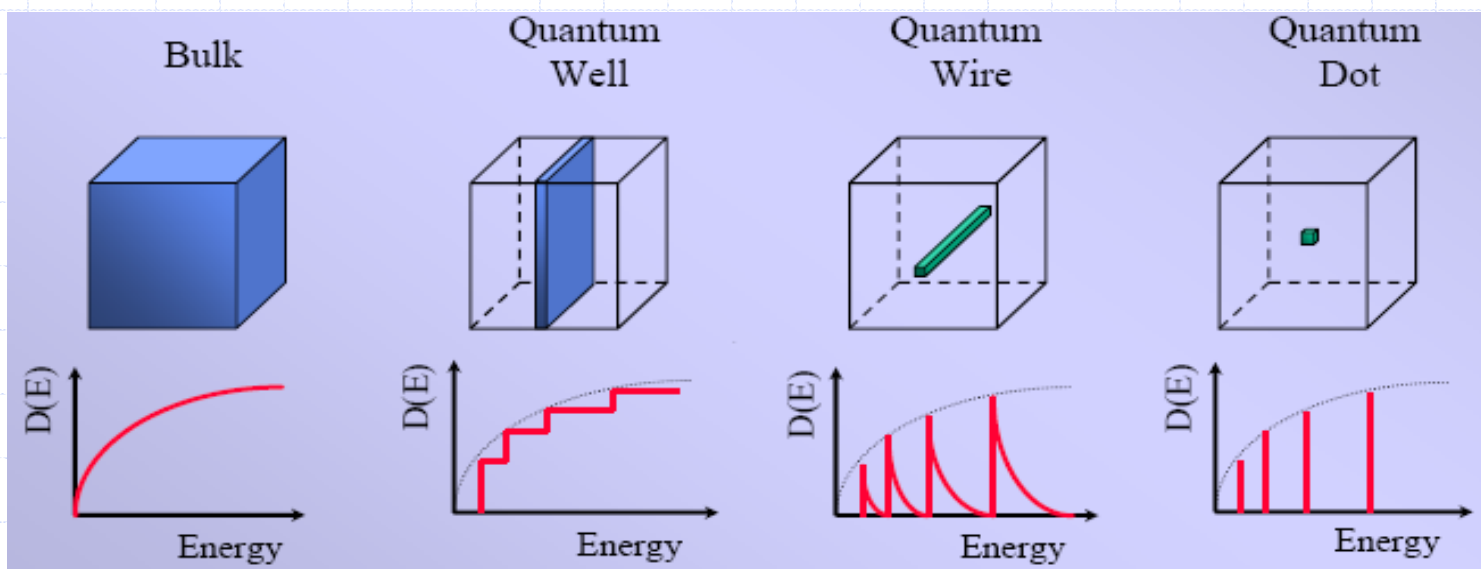
$$\text{for } L_1 = L_2 = L_3 = L$$

$$E = \frac{\pi^2 \hbar^2}{2mL^2} (n_1^2 + n_2^2 + n_3^2)$$

- States with different quantum numbers but the same energy are called degenerate

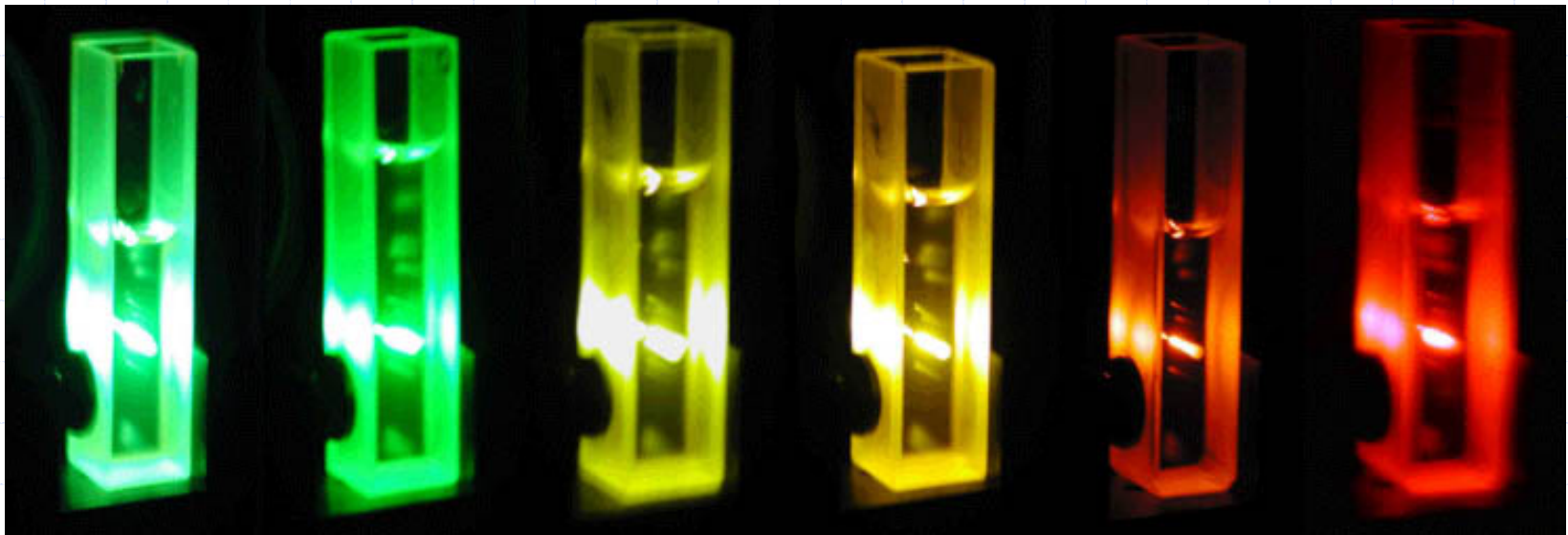
# Quantum Dots

- Quantum dots are semiconductor nanostructures that confine 1-1000 conduction electrons in all three dimensions (just like the particle in a box)



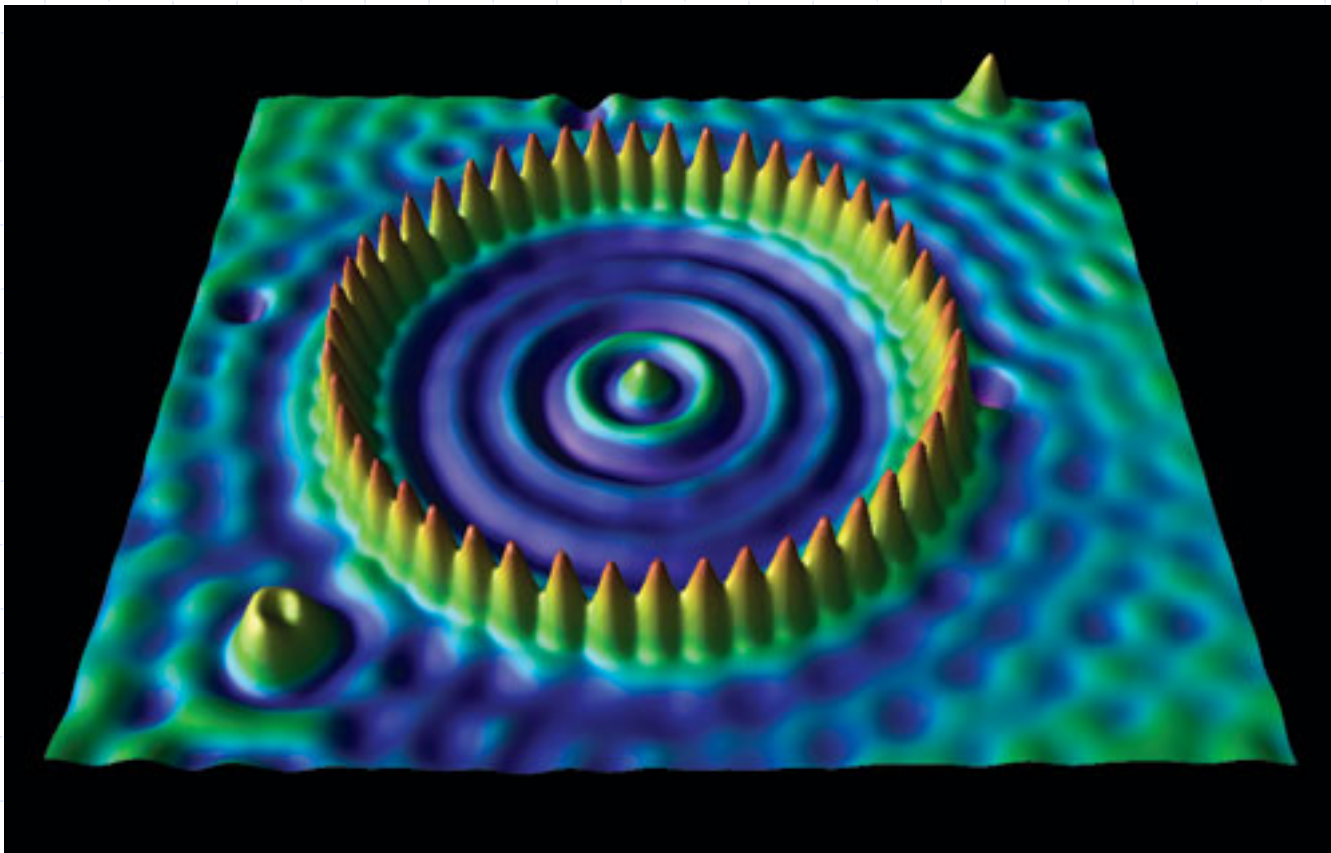
# Quantum Dots

- CdSe quantum dots in solution under exposure to UV light
- Which sample has the largest quantum dots?



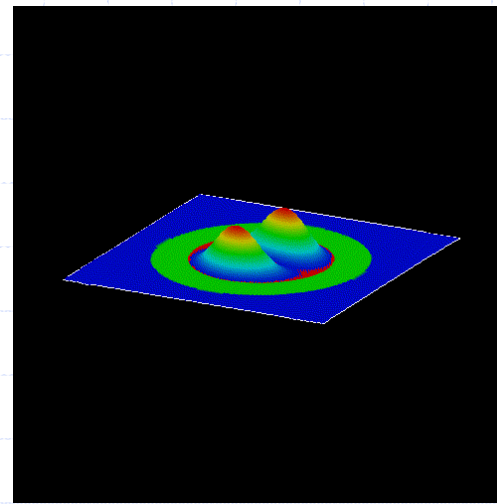
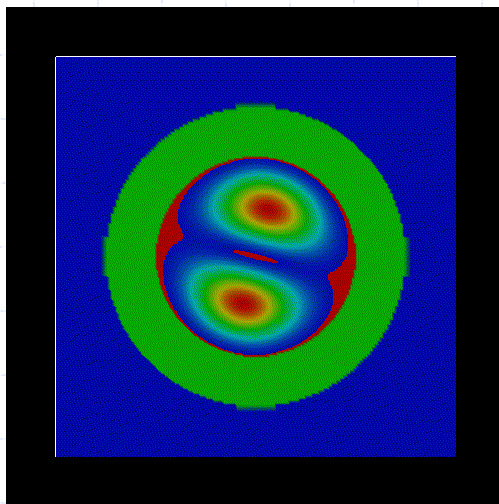
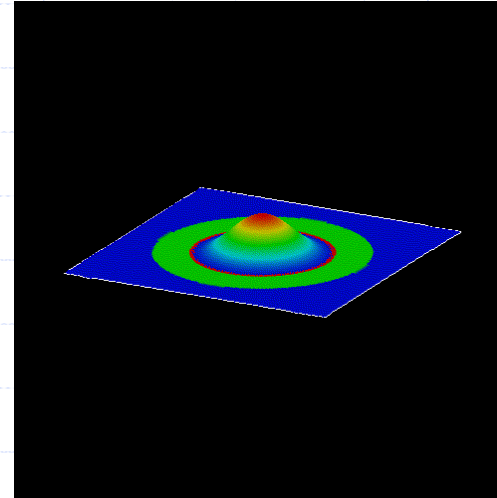
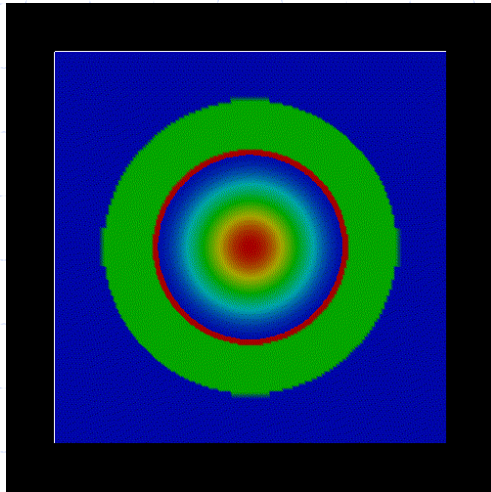
# Quantum Corrals

- Electron in a corral of iron atoms on copper



# Quantum Corrals

➤ Electron in a corral of iron atoms



# Quantum Corrals

